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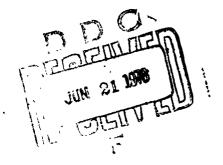
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DIFFERENTIAL-HALL IMPURITY

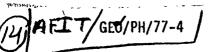
PROFILING OF GaAs, CO-IMPLANTED

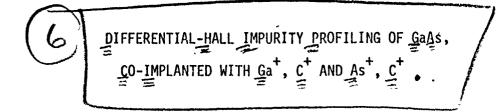
WITH Ga⁺, C⁺ AND As⁺, C⁺

THESIS

GEO/PH/77-4 Michael Stefiniw 2Lt USAF

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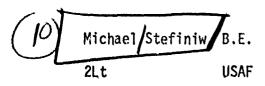
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Presented to the Faculty of the School of Engineering of the Air Force Institute of Technology

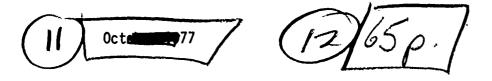
Air University

in Partial Fulfilment of the
Requirements for the Degree of
Master of Science

by



Graduate Electro-Optics



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Preface

From the start, this thesis has been most interesting and informative. It was most rewarding to do this thesis for two reasons. First, not much research has been done in dual implantations, and secondly, the results obtained suggest the possibilities which this field of work offers. Future work should be carried out on the carbon-ion for it shows the possibility of a useful dopant for submicron devices, since the diffusion is negligible up to annealing temperatures of 900°C.

I wish to thank the people who made this project possible for me.

I would particularly like to thank Jim Ehret who implanted my samples,

Charles Geesner for capping my samples as quickly as possible and

Captain David Lank for letting me use his computer profiling program

and Bok Kyoon Shin who was my lab advisor and gave me many suggestions.

Additional thanks go to my project sponsor Yoon Soo Park and faculty

advisor Dr. T. Luke.

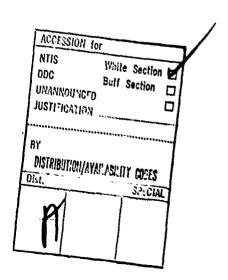


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Abstract

Co-implantation of Ga⁺ or As⁺ with C⁺ has produced p-type conduction in GaAs substrate. Sheet resistivity and Hall effect measurements were made and differential Hall effect measurements were carried out using a guarded van der Pauw system. These measurements were carried out on 10¹², 10¹³, and 10¹⁴ dosage samples annealed at 600, 700, 800 and 900°C for fifteen minutes. Doping efficiency of 87% was obtained in the 800°C annealed sample implanted with 10¹³ ins/cm² of Ga⁺ and C⁺, which is much higher than for the C⁺ alone implants. Samples annealed at 900°C, 10¹⁴ ins/cm² of Ga⁺ and C⁺ had an efficiency of 32.3%, which is also higher than the C⁺ alone implants. Carbon implants alone had a 57% efficiency when annealed at 900°C with a dose of 10¹³/cm² and 9.5% when annealed at 900°C with a dose of 10¹⁴/cm², which verifies previous work done with carbon implants.

I. Introduction

The need for devices such as laser diodes, microwave components, integrated optics, and digital IC's with multigigabit data rates has prompted research in the ion implantation field of semiconductor fabrication. The reason for using ion implantation and why so much time and effort is being placed on it, is that the doping densities, distribution and layer thickness are thought to be controlled more precisely than with diffusion or epitaxial growth.

ion implantation is a relatively new method of doping GaAs semiconductors but has been successfully used in devices with silicon substrates. It is a process by which impurity ions are accelerated by a particle accelerator and implanted into the substrate. The depth to which the impurity ions will penetrate is dependent upon the ion mass, energy of the beam and crystal orientation. After the implantation crystal lattice damage exists which must be annealed by heat treatment. This process is also used to electrically activate the impurity ions for conduction by moving them into the crystal sites. However, the annealing temperature required to effectively do the job exceeds the temperature where the GaAs surface begins to decompose. To prevent or reduce this effect the sample is encapsulated by a dielectric layer of Signs, approximately 1000A) thick. This layer is placed on the sample after implantation and before annealing. The dielectric layer is etched away when electrical measurements are performed on the sample. The guarded van der Pauw measuring system is used to acquire the crystal and Hall voltages of the samples. Electrical profiles are obtained on the 900°C annealed samples by a successive etching technique.

ANGSTROM

Statement of Problem

There has been very little research done in the area of carbon ion implanted gallium arsenide. The research that has been performed has produced results demonstrating that p-type layers can be produced in ion implanted samples. There has been no research investigating the possibility of forming an n-type layer in GaAs with carbon.

The problem to be examined in this thesis is that of forming an n-type layer in GaAs, by co-implantation of carbon ions with arsenic incoporation. The possibility of increasing the electrical activity of the carbon ions in p-type material by co-implantation with gallium is also examined. The former objective is worth the time to investigate because experimentation has shown that the group IV impurity atom tend to replace the larger of the two-sub-lattice atoms which form the compound (Ref 2:70). This would have technological value in that submicron devices of both n- or p-type layers could be formed by doping with carbon ions and controlling the GaAs stoicheometry by either Ga or As co-implantation.

Summary of Current Knowledge

No work has specifically been done in obtaining an n-type layer in GaAs using carbon ion implantation. However, all group IV elements have been successfully used in forming n-type layers, with the exception of carbon. Previously, some work was done in p-type layer implants using carbon but the doping efficiencies were so low that it was believed carbon might be useless as an impurity for ion implantation. Some research done recently along these lines has been concerned with obtaining p-type layer in GaAs using carbon ion implantation (Ref 4). The

results were a great improvement over previous work, in some cases, doping efficiencies $(n\equiv P_s/^{\phi})$ of 50% were obtained, compared with previous efficiencies of 7%. Co-implantation has been shown to enhance donor and acceptor activity in selenium and silicon implants (Ref 5,6). The problems encountered are complex but the potential of this technique warrants work in this area.

Theory

The theory behind the work performed here was developed by Hecking-bottom and Ambridge for binary compound semiconductors such as gallium arsenide (Ref 1). The reason such a theory was developed stemmed from the great difference in implanting into a element such as silicon and binary compounds. The model to represent the doping of silicon is as follows, using Kroger-Vink notation (see Appendix A):

$$F(g) \xrightarrow{X} F_{Si}^{X} \tag{1}$$

$$F_{Si}^{X} \longrightarrow F_{Si}' + h'$$
 (2)

The situation is much simpler here due to the fact that there is only one sub-lattice to build up. When an dopant is implanted in the binary compound case, only one sub-lattice is built and a compensating defect is formed (Ref 1). The situation can be described as follows:

$$F(g) = F_{AS}^{X} + V_{GA}^{X}$$
 (3)

$$V_{GA}^{X} \stackrel{:}{\longleftarrow} V_{GA}^{\prime} + h^{\bullet}$$
 (4)

which disrupts GaAs stoicheometry. The method of controlling this nonstoicheometry is to implant an equal amount of the host atom on the other sublattice when the dopant is being implanted. This is only part of the sublattice control which must be done. Time dependent non-stoicheometry arises during the annealing process. The encapsulation process is used to combat this out-diffusion.

Scope

The work performed is limited to electrical measurements and profiling using the van der Pauw system. Due to the time involved, surface measurements were taken for all the samples while only five profiles were completed. This gave some insight to the processes taking place but is in no way complete. Glow Discharge Optical Spectroscopy which would give a spatial distribution of the total impurity, was not done, due to the lack of equipment accessibility. Two such profiles, one obtained electrically and the other obtained optically, could be compared and correlated. This would give us a better understanding of the distribution of impurity ions that are used in this study.

In the following chapter, the various sections discuss sample preparation, implantation, annealing, encapsulation, differential Hall technique, and data processing. The results are presented in graphical form so that the trends of the implants can be visualized. Following this chapter, a discussion is presented on what is believed to be happening, and why. A summary, conclusion and recommendation section makes up the final portion of this study.

II. Experimental Procedure

This chapter on experimental procedure will be made up of several subsections. The sections will be covered in the following order: sample preparation, implantation procedure, encapsulation and annealing process, data acquisition and processing and finally equipment list and figure of set-up.

The gallium arsenide used in this experiment was obtained from Laser Diodes Laboratories and ordered chromium doped to make it semi-insulating. Three wafers were used approximately 2 1/2 inches in diameter and 20 mils thick. They were all from the same boule, two wafers being next to each other, while the third was from a different part of the same boule. The samples have a resistivity of 10^9 ohms and a chromium concentration of $\sim 10^{17}/\text{cm}^3$ as measured by mass-spectroscopic analysis technique.

Sample Preparation

The wafers were cut into twenty-five samples, 5mm square making maximum use of the wafer. A wafer scriber was used to cut the wafers, holding them in place by a vacuum chuck. This size was found to be the best to handle and made implanting much easier. The reason for this is that four samples could be implanted at the same time with the same dose.

The samples were then cleaned to remove any organic substance or film which might have been present on the polished surface. The sample was placed in a solution of aquasol concentrate and water in a ratio of 10:1 by volume. This soap solution, in a beaker was placed in an ultrasonic cleaner containing water filled to the same height as the soap

solution in the beaker. The sample was then allowed to clean for five minutes. After cleaning in the soap solution, the sample was rinsed with deionized water for one minute to completely remove the soap. Then the sample was rinsed with the following solvents: trichlorethylene (CL₂C:CHCL) for twenty seconds, acetone (CH₃COCH₃) for twenty seconds and finally methanol (CH30H) for twenty seconds, in that order. The sample was then blown dry by a spray of purified nitrogen gas. The surface was then checked for any organic residue and surface oxidation by dark field microscopy. If either of these were found not to be removed by the above procedure, the cleaning process was repeated until all the residue and oxidation was gone. When the surface is clean of residue, the free etch (described next) was used to clean and remove any mechanical damage which might be present. (This is necessary for coherent adhesion of the capping material.) The free etch is composed of the following chemical mixtures: $\rm H_20:30\%\ H_20_2:H_2SO_4$ in a volume ratio of 1:1:5. The chip was placed in the etch for one minute while being constantly agitated. This is done so that pits or surface deterioration will not occur. When removed from the etch the sample was immediately rinsed with deionized water for one minute to remove any acid from the surface and blown dry with an inert gas. After the above procedure is accomplished the chip was placed in the ion implantation facility to protect it from the formation of oxides or organic residue on the surface.

Implantation Procedure

The 150Kev ion accelerator used in the following work was an Accelerator Inc., Modified Implanter-I, which has been highly modified

for the work being done at the Air Force Avionics Laboratory (AFAL). The machine has two means of generating ions for implantation: either a hotcathode source or a radio frequency discharge. The hot-cathode source is used to ionize solid materials, while the radio frequency (RF) discharge is used for ionizing gases. In the present investigation, both sources were used to obtain the ions for implantation. The samples which were implanted with carbon ions (C⁺) only used the RF discharge to obtain C^+ from carbon dioxide (CO_2) gas. The samples which had dual implants had to use both sources. The first batch of dual implant samples were implanted with arsenic, obtained from arsenic metal, and with carbon, in equal dosage. The second batch of dual implants were implanted first with gallium obtained from gallium triflouride and with carbon, also in equal dosage. Three doping concentrations (10¹²/cm², $10^{13}/\text{cm}^2$ and $10^{14}/\text{cm}^2$) were used for each of the three types of materials (GaAs: C⁺, GaAs: As⁺, C⁺ and GaAs: Ga⁺, C⁺) making nine different types of samples in all. The energy of the implants were 60Kev for the C⁺ and 120Kev for the gallium and arsenic ions. These were not the energies which were desired to implant the ions but they were only what the accelerator could handle. It would have been preferred to implant the carbon ion at 120Kev and gallium and arsenic ions at 700Kev, so that the two distributions of implants would overlap more completely and minimize surface erosion during implantation. The energies used did show some overlapping of distributions, but not as much as was desired.

Encapsulation and Annealing Conditions

After the implantation of the ions, the samples were encapsulated

with silicon nitride (Si_3N_4) a material which has been well documented in the literature to be superior to silicon dioxide (SiO_2) for GaAs encapsulation((Ref 4). The cap was used in this thesis to prevent gallium from out-diffusing, which occurs at temperatures of 600°C (Ref 4).

All the samples were coated with approximately 1000\AA of Si_3N_4 , as determined by the color change which takes place on the surface of the sample during the encapsulation process (Appendix C). This cap can also be placed on the surface prior to implantation so as to control the position and shape of the distribution.

The pyrolytic reactor used in this thesis for encapsulation was designed based on a MIT Lincoln Laboratories set-up (Ref 3:26). It is possible to encapsulate samples with two different types of films using this system, SiO_2 or Si_3N_4 . For this experiment, Si_3N_4 film was deposited on the sample by using specific amounts of nitrogen (N_2) , ammonia (NH_3) and silane (SiH_4) mixed in a glass bell chamber. The chamber was first evacuated to remove impurities from the system, after which it was purged with nitrogen. The sample rested on a carbon strip whose temperature was controlled, thus allowing the reaction which is necessary for the deposition of the film. As the chamber was being purged with nitrogen, the sample was brought to 200°C, helping to prevent any silicon dioxide from forming on the surface. Formation of silicon dioxide would prevent coherent adhesion of the Si N film. After purging for a few minutes, controlled amounts of 5% saline and liquid ammonia were introduced which had been previously diluted with equal amounts of nitrogen. These mixtures were added to the chamber from

above where they were allowed to mix for a few seconds. The temperature was then brought up quickly to 740°C for thirty seconds (for 1000Å thick film), during which time the gas mixtures reacted to form a film on the sample which could be observed by the change in color of the semiconductor surface. After thirty seconds the temperature was brought down immediately to 200°C, during which time the chamber was purged of the gas mixture using nitrogen. The film was then checked for its true thickness using a Gaetner Scientific ellipsometer. This verified the thickness used on the samples in this thesis of approximately 1000Å-1200Å. The flow rates of saline, ammonia and nitrogen, the carbon strip temperature during deposition, and the time allowed for deposition are documented in Appendix B.

After the $\mathrm{Si_3N_4}$ was deposited, the samples were annealed at 600°C, 700°C, 800°C and 900°C (15 minutes at each temperature) in a flowing hydrogen ambient. Three samples of the same implants but of different doses were all annealed at once to save time. When the samples were annealed, a tub of gallium was placed with them so as to help combat the out-diffusion of gallium (Fig. 1). To minimize thermal shock, the

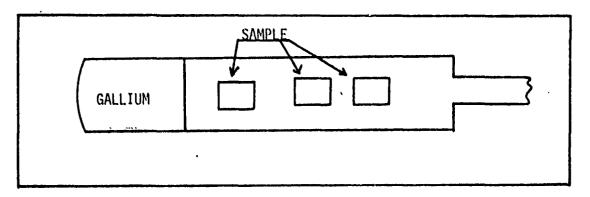


Fig. 1. Set-up for annealing samples

samples to be annealed were placed in the holder outside the quartz tube furnace, as it was brought up to temperature. The hydrogen gas was first run through a hydrogen purifier before flowing into the furnace. The hydrogen flows through the furnace at 226.6 cubic centimeters per minute during the warm up period, annealing process and cool down. After annealing, the samples were pulled from the heating area but not removed from the furnace, thereby allowing them to cool down to room temperature. During cool down, hydrogen still flows through the furnace and over the samples, at the same time being bled off by burning on a platinum wire. After cooling down, the samples were removed from the furnace and were ready to be profiled once the cap was removed and contacts were attached to the samples.

The cap was removed using a 48% reagent hydrofluoric acid (HF), for usually ten minutes or as long as required. Then the sample was washed by flowing deionized water over it for one minute (to remove the acid from the surface) and blown dry using nitrogen gas.

Contacts were then ultrasonically soldered to the four corners of the sample using pure indium. After soldering, the contacts were checked with a curve tracer to make sure t'ey were ohmic (Fig. 2). If they were not ohmic, the contacts would be resoldered in an attempt to correct the problem. If, after resoldering the contacts were still rectifying they could be sintered by heating the sample to 450°C. This would reduce the strain in the crystal and the contacts would be more ohmic in nature as a result of the alloying process.

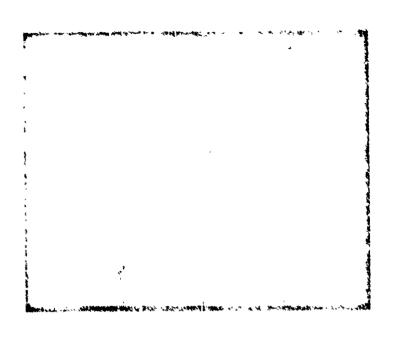


Fig. 2. I-V curve for sample contact

Data Acquisition and Processing

The system used for data acquisition was a van der Pauw measurement system composed of guarded unity-gain amplifiers (Fig. 3) which provide an input impedance of 10¹⁴ ohms. This system was used at the Aerospace Research Laboratory and designed by B. K. Shin. It has a variety of measurement copabilities such as a guarded Hall and van der Pauw configurations and continuous temperature changes to 4°K using the Air Product Helitran.

The van der Pauw method was the only one used in this thesis to acquire the data. There are advantages in using this method in that the samples were small and only four contacts were required on the periphery of an arbitrarily shaped sample. When using the van der Pauw method, it is required to measure voltage across two leads while current

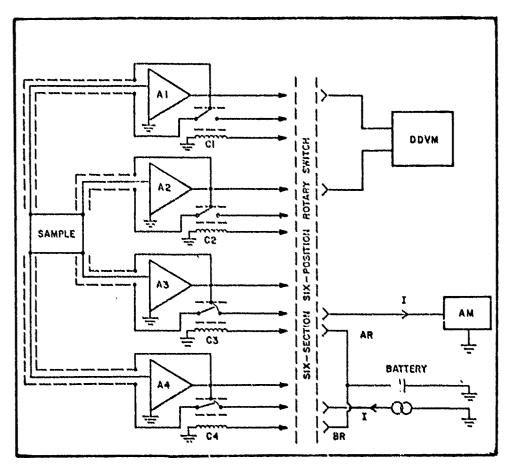


Fig. 3. System diagram of guarded van der Pauw

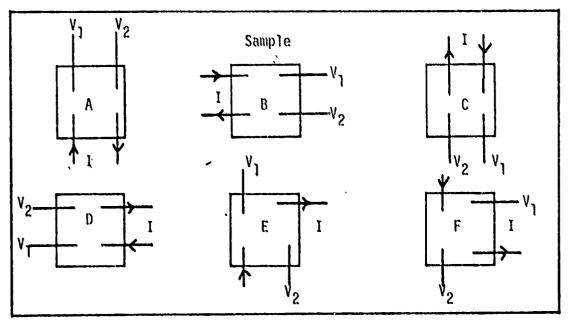


Fig. 4. Sample configurations

is passed through the opposite leads as shown in Fig. 4 (a-d). Voltage readings were made by interchanging the leads to the sample. This was done by a six-position rotary switch which is marked the same as the configurations in Fig. 4. The last two configurations (e & f) were used to take Hall voltage readings when a magnetic field is applied prependicula— to the sample. A total of eight voltage measurements were made on configurations a-d by reversing the current direction on each configuration. This way, after averaging all the crystal voltages, a more realistic value of the samples resistivity would be arrived at. The Hall voltage was also determined more realistically by first applying the field in one direction and reversing the current direction, then reversing the field and reversing the current direction again. There was then a total of eight Hall voltages from which an average was used to arrive at a Hall coefficient.

From the average crystal voltage an effective sheet resistivity was determined using the following equation:

$$f_{s} = \frac{\pi}{n^{2}} \frac{\Delta^{V_{c}}}{I_{c}} f \qquad (\Omega/\Box)$$
 (5)

where P_s = sheet resistivity

 ΔV_c = average crystal voltage

I_c = applied current

and f = dimensionless correction factor.

The Hall voltage was used to arrive at a Hall coefficient from the following equation:

$$R_s = 10^8 \left(\frac{\Delta V_s}{I_c B}\right) \qquad (cm^2/coul) \qquad (6)$$

where $R_s = Hall$ coefficient

 ΔV_{s} = Hall voltage

 $I_c = applied current$

and B = applied magnetic field.

 $\rm R_{\rm S}$ was then used to calculate an effective sheet concentration, $\rm N_{\rm S}$, and mobility, $^{\rm H}$ eff:

$$N_s = (qR_s)^{-1}$$
 (cm⁻²) (7)

where q = electron charge

$$\mu_{\text{eff}} = \frac{R_{\text{S}}}{P_{\text{S}}} \qquad (cm^2/V-\text{sec}) \qquad (8)$$

The above equations are just averages depending upon the distribution in depth of the carrier concentration and the implantation factors. However, an electrical profile can show what fraction of the implanted atoms become substitutional, as well as the spatial location of the active atoms with respect to LSS theory (see Appendix F). The electrical profile was obtained by removing thin layers of the sample at a controlled rate. As each layer was removed, sheet resistivity and Hall coefficient measurements were taken and the number of carriers and their mobility was determined from the following equations (Ref 11):

$$\frac{(R_s)_i}{(f_s)_i^2} - \frac{(R_s)_{i+1}}{((f_s)_{i+1})^2} = en_i^{\mu} i^2 d_i . \tag{9}$$

and

$$\frac{1}{(P_s)_i} - \frac{1}{(P_s)_{i+1}} = en_i \mu_i d_i$$
 (10)

From equation (9) and (10) if follows that

$$\mu_{i} = \frac{(R_{s}/f_{s}^{2})i}{(1/f_{s})_{i}}$$
 (11)

and

$$n_i = \frac{(1/P_s)_i}{ed_i u_i}$$
 (12)

where d_i is the thickness of the ith layer removed. The carrier concentration, $N_s(cm^{-2})$, is obtained from n_id_i .

The degree of electrical compensation is defined as the ratio of donor to acceptor concentrations, $K = N_d/N_a$. It was not necessary to compute the values of N_d and N_a to find the compensation level. The only value that had to be determined was the concentration of total ionized impurities, P_I . This was done by using the well known Brooks-Herring relationship (Ref 11).

$$\mu_{\rm I} = \frac{3h^3k^2p}{16^2e^3m^{*2}P_{\rm I}} = \frac{1}{\ln(1+f)-f/(1+f)}$$
 (13)

where $f = \frac{1}{2} \left(\frac{h}{e} \right)^2 \left(\frac{k}{m^*} \right) \left(\frac{3p}{8} \right) \frac{1}{3}$

h = Plank's constant

k = dielectric constant

e = electron charge

p = carrier concentration

 P_{T} = concentration of total ionized impurities

m* = effective hole mass

To be able to use the above equation, the mobility due to the ionized impurities (μ_1) must be known. This can be determined from the following equation:

$$\frac{1}{\mu} = \frac{1}{\mu_{\rm L}} + \frac{1}{\mu_{\rm L}} \tag{14}$$

assuming a value of 450 cm 2 /V-sec (p-type samples) for lattice mobility (μ_L).

In order to calculate the compensation, K, the following two equations are used:

$$P_{I} = N_{d} + N_{a} \tag{15}$$

$$p = N_d - N_a . (16)$$

These equations can be manipulated to give the following:

$$P_{T} - p = 2N_{d} \tag{17}$$

$$P_{T} + p = 2N_{a}$$
, (18)

from which follows:

$$K = \frac{P_I - p}{P_I + p} \tag{19}$$

Using equations (13), (14) and (19) a series of curves can be formed by choosing a vaule for K, such that $0 \le K \le 1$, and a value for p, see Fig. 5.

The removal of these layers was obtained by etching the sample with a solution of H₂SO₄: 30% H₂O₂: H₂O in a ratio of 1:1:50, kept at O°C. The etch removed thin layers of the sample at a predicted rate of 250 Å/min as determined by a Sloan Dektak microtopography on unimplanted sample. The contacts were protected at the same time to prevent degradation of the ohmic characteristics. This was done by covering the electrodes with black wax which has been mixed with trichlorethylene to a paint-like consistency. After profiling, the black wax was removed by washing the sample with trichlorethylene, thus revealing the contacts. Then the sample was taken to the Dektak to verify whether the etch was still etching at the determined rate. It was found that the etch actually proceeded at a slower than predicted

rate of 144 Å/min. After discovering this phenomena each sample was measured on the Dektak after profiling. The results showed that the etch rate was the same for the different implants and dosage. Therefore the implantation of the ions into the substrate must cause some change in the crystal for it to react in such a manner.

III. Experimental Results

Isochronal Annealing

This section deals mainly with temperature dependent parameters, obtained from surface measurements. The three different types of implants (GaAs:C⁺, GaAs:Ga⁺, C⁺, GaAs: As⁺, C⁺) discussed individually and compared with each other to see if there was any verification of the predicted results. In addition an encapsulation study is included due to the problems which have been plaguing the process. This additional study helped to distinguish the effects of implantation versus the effects of encapsulation. The only data that was obtained was that of resistivity. No carrier concentration or mobility data was obtainable due to the Hall voltage being buried in the noise.

For the carrier concentration and mobility versus dose graphs, for all three implants, the numbered points 6, 7, 8, and 9 stand for 600°C, 700°C, 800°C, and 900°C, temperatures at which the samples were annealed.

GaAs:C+

Annealing at the temperatures stated produced electrical activity in samples having a dose of $10^{13}\text{C}^+/\text{cm}^2$ and $10^{14}\text{C}^+/\text{cm}^2$ as shown in Fig. 6. For these two doses, it can be seen that the effective sheet resistivity is in the same order of magnitude for all annealing temperatures (T_A) . The sample implanted with a dose of $10^{12}\text{C}^+/\text{cm}^2$ shows a very high resistivity, the same as that of the sample which was just encapsulated and annealed. It decreases sharply as does the encapsulated sample at 700°C and 800°C. For the temperature range of 800-900°C, the

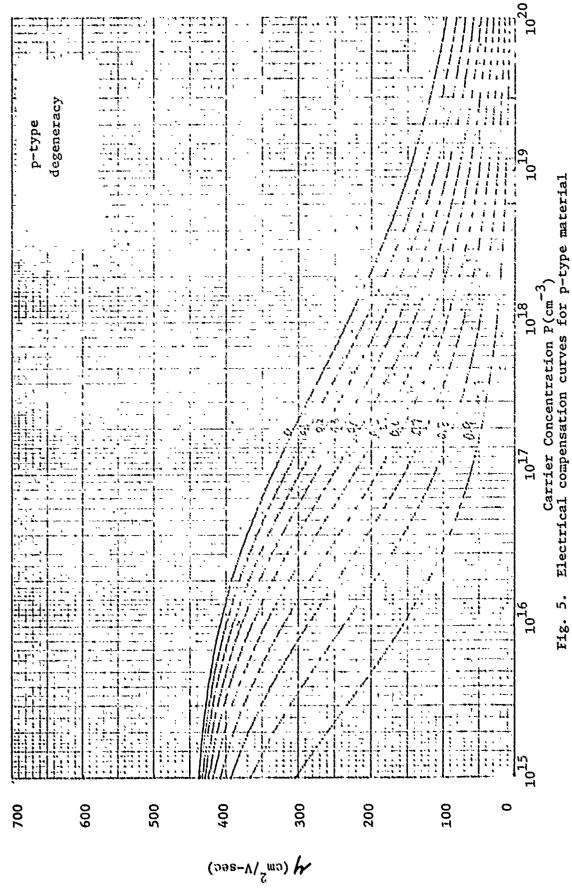


Fig. 5.

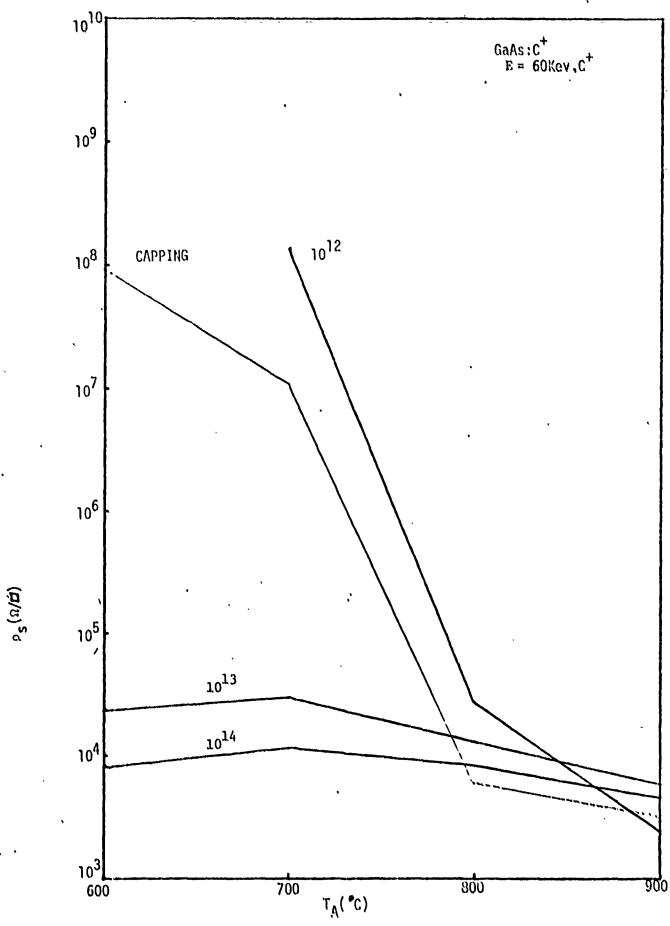


Fig. 6. Effective sheet resistivity (f_s) vs. annealing temp. (T_Λ) for C

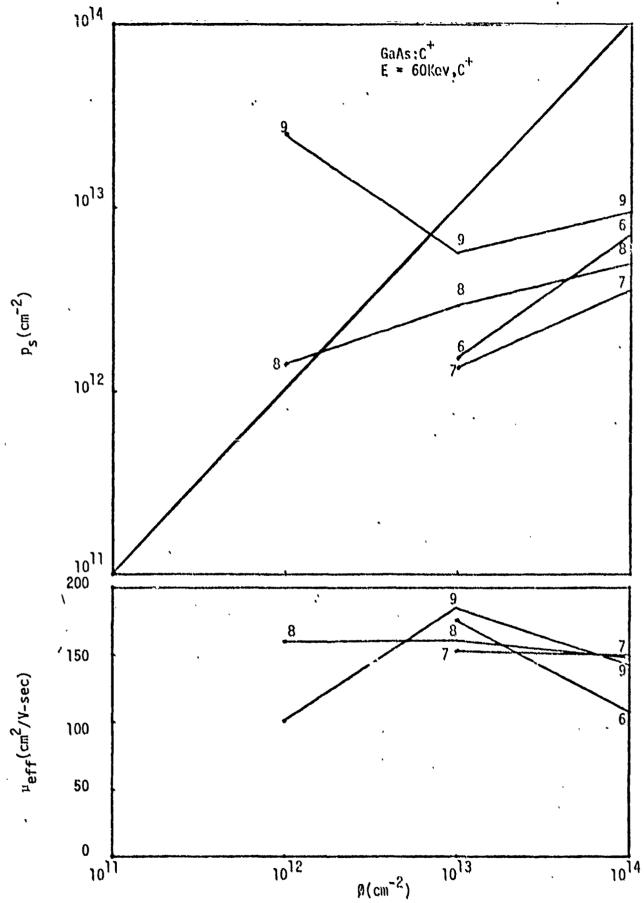


Fig. 7. Dependence of sheet hole concentration (p_s) and effective mobility upon C dose (β)

resistivity is within the same order of magnitude for the different dosages and encapsulation. It can be said that there is some observed implantation effect in the $10^{13} \text{C}^+/\text{cm}^2$ and $10^{14} \text{C}^+/\text{cm}^2$ dosages at 600°C and 700°C samples. It is difficult to say this for the 800°C and 900°C range due to the gathering of all the curves at these temperatures.

Fig. 7 shows the annealing characteristics of effective sheet carrier concentration (P_s) and effective Hall mobility (µeff) for different temperatures. For $10^{12}\text{C}^+/\text{cm}^2$, carrier concentration and mobility were only obtainable for 800°C and 900°C. The carrier concentration for 800°C and 900°C increases with temperature but was higher than the implanted dosage. Also observed was the effect that the concentration decreased between 600°C and 700°C, and then increased between 700°C and 900°C for both the $10^{13}\text{C}^+/\text{cm}^2$ and $10^{14}\text{C}^+/\text{cm}^2$ samples. Although this trend was observed the concentration for $10^{14}\text{C}^+/\text{cm}^2$ at all annealing temperatures is higher than $10^{13}\text{C}^+/\text{cm}^2$ samples.

The doping efficiencies (n = P_s/p) were also determined in this study from the diagonal line drawn across the graphs. The results were compared with previous work done in this area by Sansbury, Harris and Shin (see Table I). For a dose of $10^{13}C^+/cm^2$, $900^{\circ}C$ annealed, an efficiency of 57% was obtained while an efficiency of 9.5% was seen for a dose of $10^{14}C^+/cm^2$ annealed at $900^{\circ}C$. These values agree very well with Shin's findings even though an implantation energy of 60Kev was used. The low values obtained in previous studies were probably due to the low annealing temperatures and encapsulation methods used. In this study the same annealing temperatures and encapsulation procedures were used as by Shin (Ref 4).

Table I

Electrical Characteristics for Various Carbon ion Implantation (Ref 4)

Ø(cm ⁻²)	E(Kev)	T(°C)	$\mu_{\rm eff}({\rm cm}^2/{ m V-sec})$	n(%)	Capping	Refferences
1013	70	700	29	1.9	510,	Sansbury
1014		700	242	0.95	S ₁₀₂	Sansbury
1015		700	282	0.064	Si0 ₂	Sansbury
2×10^{14}		800	240	1.5	S10 ₂	Harris
2×10^{14}		800	180	7.01	Si0 ₂	Harris
1013	120	006	218	50.0	Si ₃ N ₄	Shin
1014	120	006	224	12.0	Si 344	Shin
1013	09	006	185	57.0	Si3N4	This work
1014	09	006	143	9.5	Si 3N4	This work
1013	120(Ga ⁺)	800	125	87,0	Si3N4	This work
1014	120(Ga ₊) 60(C ⁺)	006	116	32.0	Sī3N4	This work

¹Sample implanted at 77°K

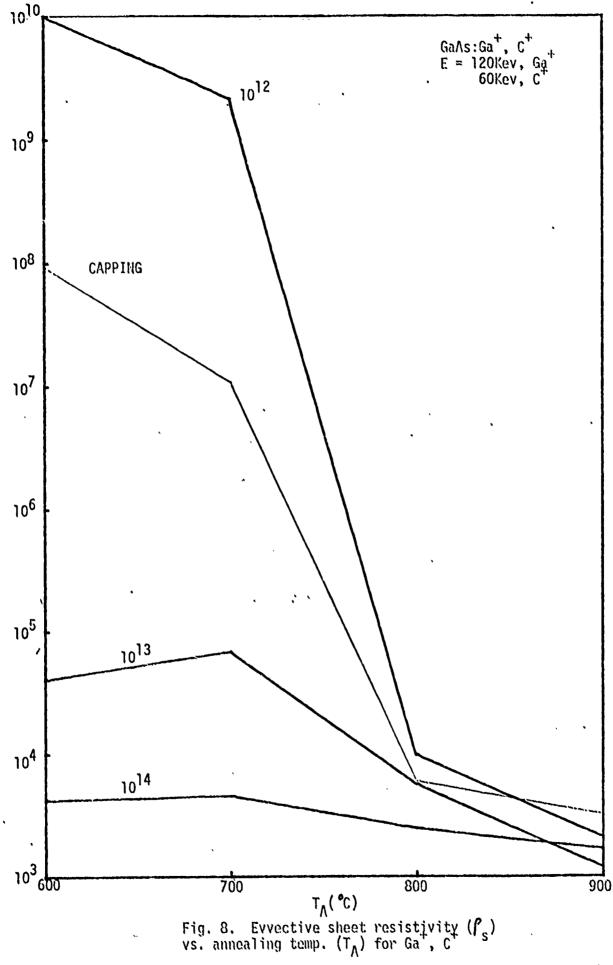
From the Hall mobility we can also see the effect that a dose of $10^{13}\text{C}^+/\text{cm}^2$ annealed at 900°C is more effective than a dose of $10^{14}\text{C}^+/\text{cm}^2$. The mobility is higher for $10^{13}\text{C}^+/\text{cm}^2$ dose than $10^{14}\text{C}^+/\text{cm}^2$ dose for all temperatures but lower than obtained in previous studies. This higher mobility could result from the encapsulation problem.

GaAs:Ga+C+

The effective sheet resistivity (f_s) for GaAs: Ga,C+ at the different annealing temperatures (Fig. 8) looks very much like what was observed for GaAs:C+. The encapsulation and 10^{12} Ga+, C+/cm² samples were high, while 10^{13} Ga+, C+/cm² and 10^{14} Ga+, C+/cm² dose samples were in the 10^4 and 10^5 ohms per square (Ω/\Box) range. Another observation made was that all the dosages in the 800°C to 900°C range had lower resistivities than the GaAs:C+ samples in the same range.

More observations can be made by looking at Fig. 9 which shows the effective sheet hole concentration and effective Hall mobility as a function of implanted dose (\emptyset). First, it is evident that for every temperature the hole concentration is greater for 10^{14}Ga^+ , C^+/cm^2 than 10^{13}Ga^+ , C^+/cm^2 . The most promising fact is that at 900°C for 10^{14}Ga^+ , C^+/cm^2 the hole concentration is $3.23 \times 10^{13}/\text{cm}^3$ which is an efficiency of 32%, a noticable improvement over just carbon implants. However for the 10^{13}Ga^+ , C^+/cm^2 dose at 800°C we have an efficiency of 87%, which is also a large increase over the same dose and temperature for just carbon implants.

The effective Hall mobility also showed a trend, in that at the 10^{14} Ga⁺, C⁺/cm² dose the carbon implants alone were much higher than



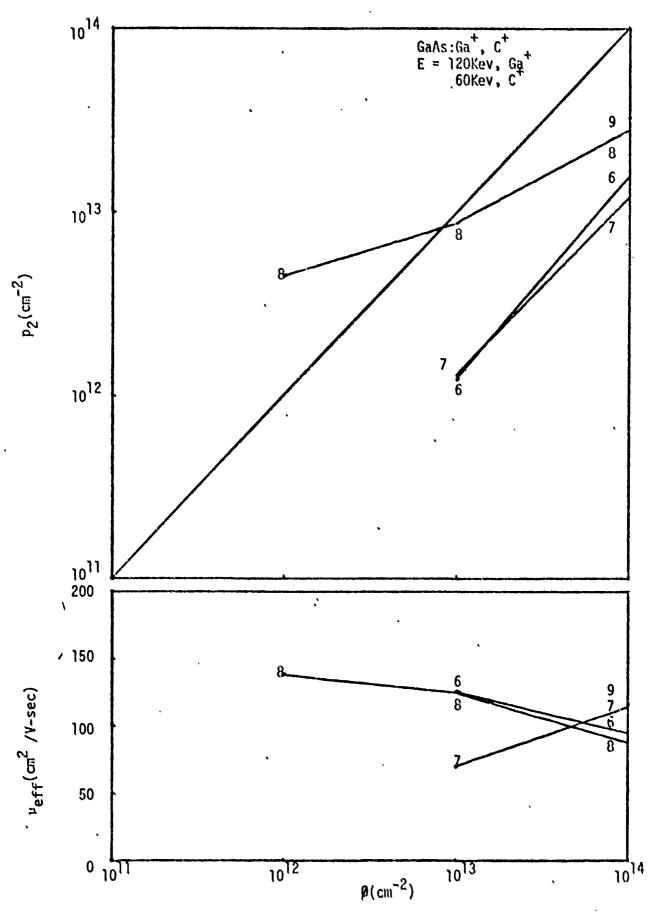


Fig. 9. Dependence of sheet hole concentration (p_s) and effective mobility upon Ga , C dose (\emptyset)

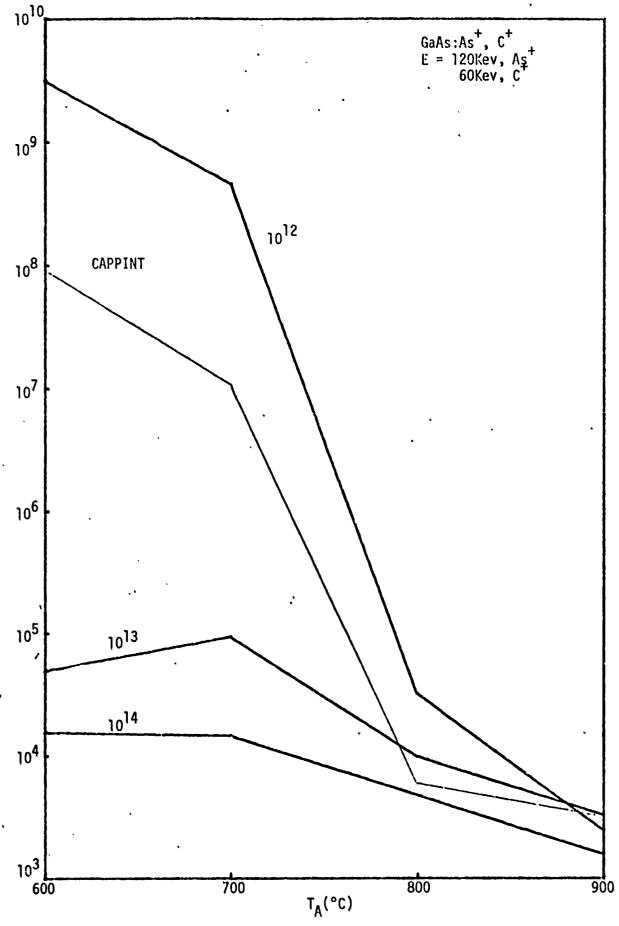
the Ga^+ , C^+ implants. In addition, there was an observed decrease in mobility from 10^{13} to 10^{14} doses in both cases for all temperatures except at $700^{\circ}\mathrm{C}$.

GaAs:As⁺, C⁺

The effect sheet resistivity Fig. 10 shows the same trend as that of the C⁺ and Ga⁺, C⁺ implants. Only a small difference can be seen in the 10¹⁴ dose where the sample has a little higher resistivity in the 600°C to 700°C temperature range. The graph also shows the same trend as the others, that is, from 800°C to 900°C they all come together. This makes it very hard to determine if this effect resulted from encapsulation or implantation effect.

Looking at Fig. 11 we still observe the strange phenomena of the concentration being larger than the implanted dose this time for a 10^{13}As^+ , C^+/cm^2 dose at 900°C annealed. This characteristic may also have been observed for Ga^+ , C^+ implants if it were possible to measure the Hall voltage. Also observed was that for 600°C and 700°C annealing temperatures at 10^{13}As^+ , C^+/cm^2 and 10^{14}As^+ , C^+/cm^2 dosages the concentration was lower than for all the other implanted samples. Also, the efficiency for the sample annealed at 900°C at a dose of 10^{14}As^+ , C^+/cm^2 was 25%, which is lower than the Ga^+ , C^+ implants but much higher than the C^+ implant.

The effective Hall mobility plot shows that at 10^{14} dose for all temperatures the mobility is in the same region as for C^+ implants but higher than for the Ga^+ , C^+ implants. At the same time, the 10^{13} dose of As^+ , C^+ implants for all temperatures had mobilities which were lower than the C^+ implants and in the same region as the Ga^+ , C^+ implants.



ر*ه/ه*) s

Fig. 10. Effective sheet resistivity (f_s) vs. annealing temp. (T_A) for As , C

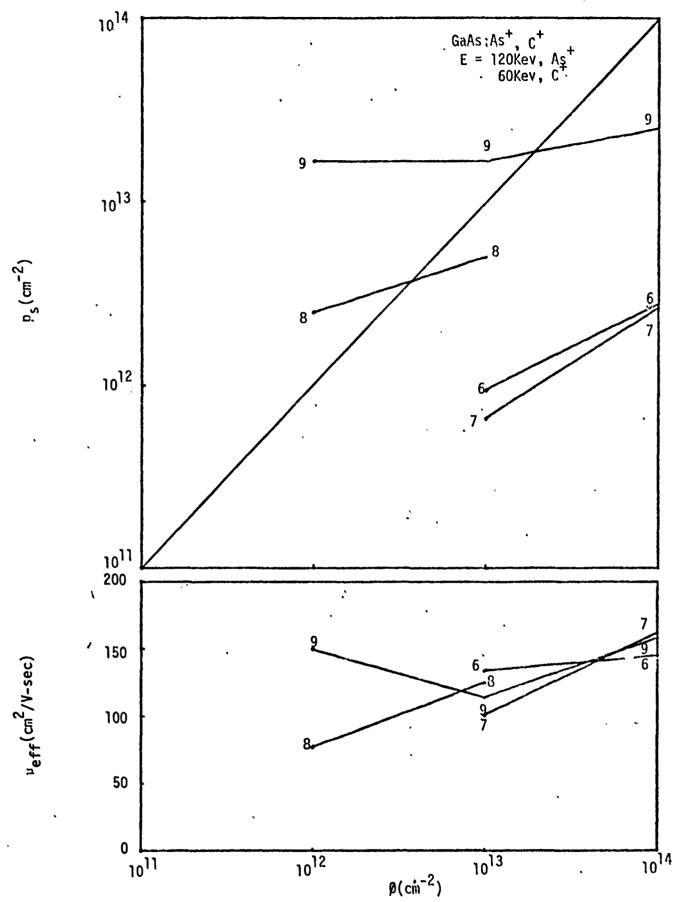


Fig. 11. Dependence of sheet hole concentration (p_s) and effective mobility upon Λs , C dose (0)

GaAs:Ar+

Due to problems in resolving whether or not the different trends in the data were due to implantation, damage or encapsulation argon implants were done. These implants would help in determing whether or not we were seeing implantation or damage effects, there was no experiment which could have been performed to separate the effects of encapsulation or implantation. The samples were prepared in the same matter as the C⁺, Ga⁺, As⁺ samples, so the encapsulation effect would also be present. Only 600°C and 900°C annealed samples with 10^{12} , 10^{13} and 10^{14} doses of argon were made and encapsulated with 1000Å of Si_3N_4 . Surface measurements were taken and resistivity, mobility and concentration were calculated, the results follow.

Table II GaAs:Ar⁺

T _A (°C)	Ø(cm ⁻²)	f _s (n/d)	μ (cm ² /V-sec)	P _s (cm ⁻³)
600°C	1012 1013 1014	5.8 x 10 ⁷ 2.8 x 10 ⁷		
900°C	1012 1013 1014	5.2 x 10 ⁰ 7163 11000 23600	154 120 91.5	5.7 x 10 ¹² 4.7 x 10 ¹² 3.0 x 10 ¹²

We can see from the 600°C annealed samples that the resistivity stays fairly constant for the different dosages. While for the 900°C annealed samples the resistivity increased as the dose increased which can be taken as an increase in damage to the crystal. This damage effect can also be seen in the mobility and carrier concentration data.

Electrical Profiles

This section of the results will deal mainly with the electrical profiles of GaAs: As^+ , C^+ (\emptyset = 10^{12} , T_A = 900° C), GaAs: As^+ , C^+ (\emptyset = 10^{13} , T_A = 900° C), GaAs: As^+ , C^+ (\emptyset = 10^{14} , T_A = 900° C), and GaAs: Gs^+ , C^+ (\emptyset = 10^{14} , T_A = 900° C) which LSS profiles of As^+ and Ga^+ at 120Kev and C^+ at 60Kev. Two other profiles are also plotted on the same graph for each, those of V_{GA} and V_{AS} vacancies due to the experimental work of Chiang and Pearson (Ref 10). Also each plot, beneath the profile, has Brooks-Herring mobility compared to the mobility determined experimentally. Using the above information and bulk hole concentration, the electrical compensation coefficient, K, could be determined from the curves (see Figure 5).

When first comparing all the profiles, it is surprising to observe the similarities between them. The value of hole concentration is a maximum near the surface of about 10^{18} and 10^{19}cm^{-3} , and slowly varies with depth, x. The most interesting one is the GaAs:As⁺, C⁺ (\emptyset = 10^{12} , T_A = 900° C), sample where it is observed that the experimentally determined profile is actually higher than the LSS profiles which might be anticipated from the effective sheet carrier concentration measurement. The reason for this phenomena is not very definite but after looking at the Hall voltages it was observed that most of the readings were less than 100 microvolts, voltages at which error is probable.

Out of all the samples, the most interesting to compare are $GaAs:As^+$, C^+ (Fig. 14) and $GaAs:Ga^+$, C^+ (Fig. 15). The maximum concentration of $GaAs:As^+$, C^+ almost reaches $10^{19}cm^{-3}$ near the surface, which is the highest compared to the rest. $GaAs:Ga^+$, C^+ follows the

LSS profile better than $GaAs:As^+$, C^+ , which deviates from it near the tail and behaves as the gallium vacancies (V_{GA}) profile. The same result might have been observed in the $GaAs:Ga^+$, C^+ sample if it were etched a little further. Looking at the mobility curves we see that for $GaAs:Ga^+$, C^+ the mobility is more constant with depth where as for $GaAs:As^+$, C^+ the mobility varies quite a bit in the implant region, and starts to smooth out, as if it was due to the V_{GA} . While looking at the electrical compensation curves $GaAs:Ga^+$, C^+ shows a much better K near the surface and increasing with depth which is logical. The $GaAs:As^+$, C^+ sample shows a compensation curve which is virtually invarient with depth, neglecting the first two points which are due to very small Hall voltages.

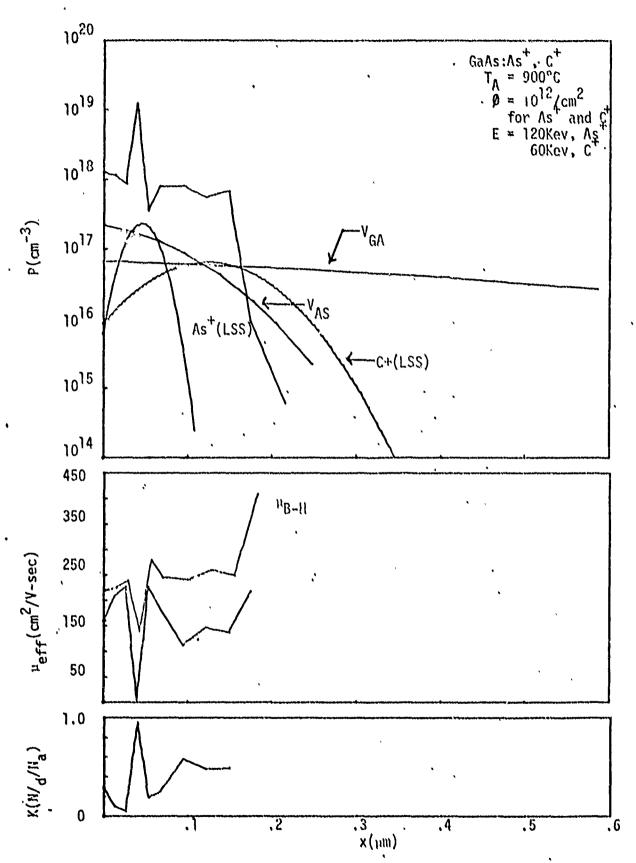


Fig. 12. Electrical profiles of carrier concentration (P), mobility (ν_{eff}) and electrical compensation factor (K) with depth (x)

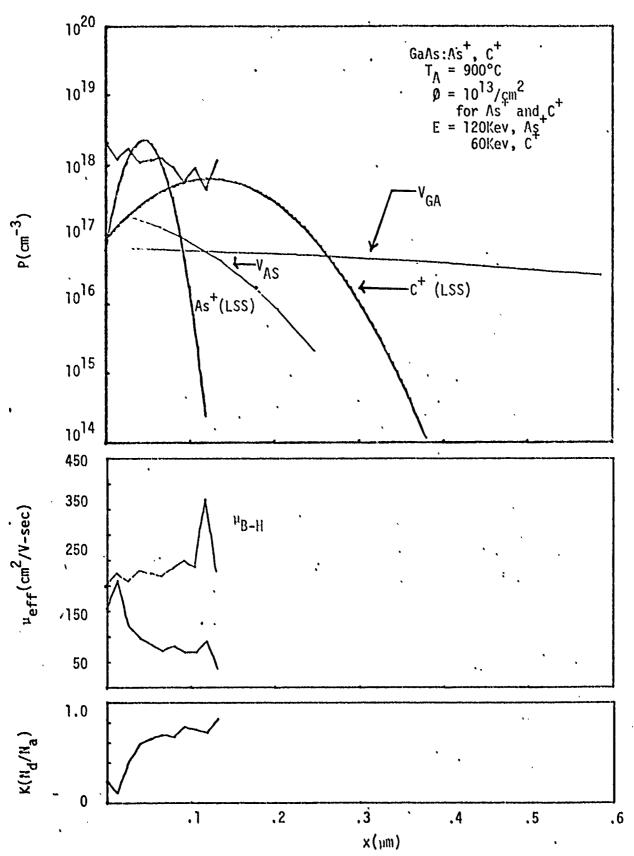
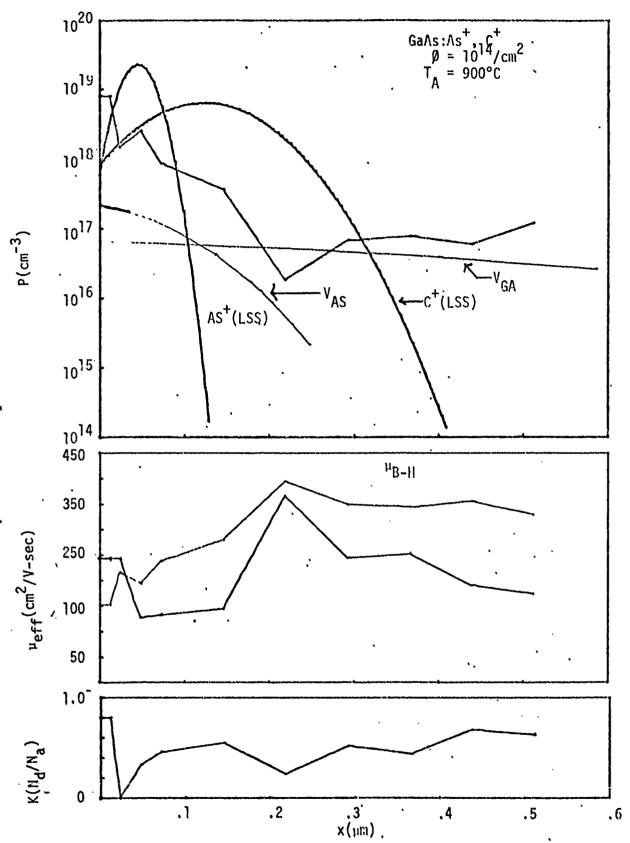


Fig. 13. Electrical profiles of carrier concentration (P), mobility ($\mu_{\mbox{eff}}$) and electrical compensation factor (K) with depth (x)

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Fig. 14. Electrical profiles of carrier concentration (P), mobility ($\mu_{\mbox{eff}}$) and electrical compensation factor (K) with depth (x)

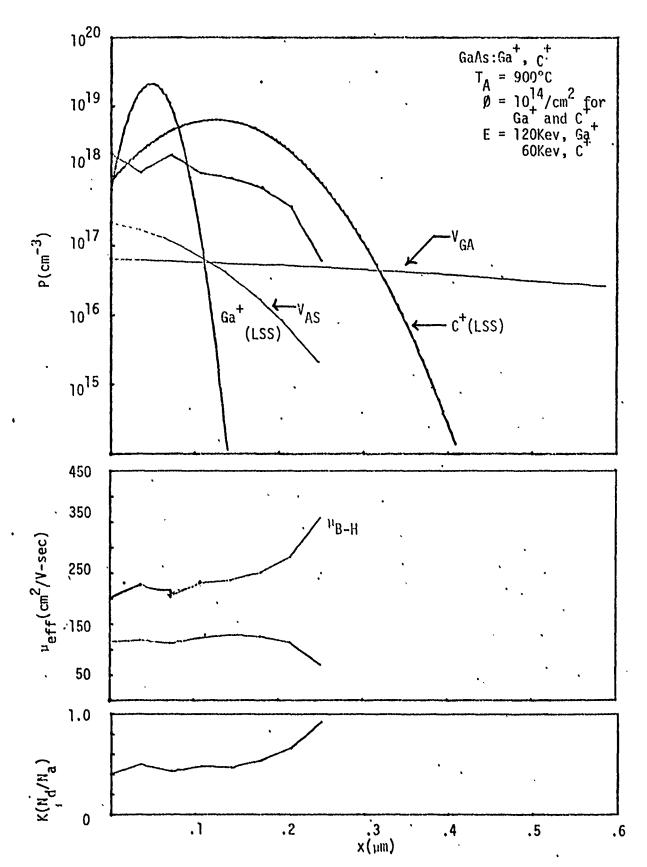


Fig. 15. Electrical profiles of carrier concentration (P), mobility (μ_{eff}) and electrical compensation factor (K) with depth (x)

IV. <u>Discussion of Experimental Results</u>

From the data acquired on isochronal annealing two features can be seen immediately. For all the different types of implants, at annealing temperatures of 600°C and 700°C, the resistivities for the 10^{12} samples are very high, higher than just encapsulated samples (no implants). As data from GaAs: Ar implants indicated that the resistivity is high. However, 10^{13} and 10^{14} dose samples for all implants show low resistivity. Such a difference in resistivity with dosage suggest there is a strong dose dependence with annealing temperature. In fact, it is so dose dependent that no matter what the implants may be, this effect was observed. At the higher annealing temperatures the resistivities are all the same order of magnitude, suggesting that the thermal effect predominates the resistivity. At these temperatures the encapsulation (with which there has been some problem) could act as a sink, drawing out the Ga and As, and causing vacancies to be formed. This possibility may also account for the observations made with the 10¹² dose samples. annealed at 800°C and 900°C which show a doping efficiency of greater than 100%. The Hall voltages measured on these samples were less than 100 microvolts and very unstable which is subject to error. The crystal voltage (which was higher and more in the range of accuracy of the equipment) also displayed a strange phenomena. Crystal voltages from one configuration to the next changed drastically, sometimes by a factor of six, indicating there was severe inhomogeneity, as can be seen in Fig. 16 by checking for ohmic contacts in different configurations.

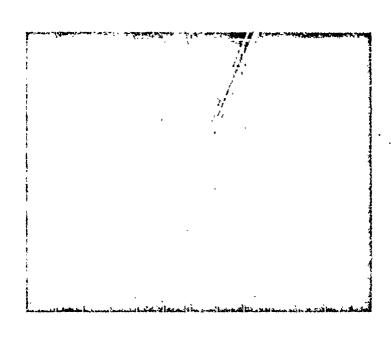


Fig. 16. I-V curves for different sample configurations

The first intention of this study was to reproduce the results of C⁺ implantation by Shin and to increase the efficiency of C⁺ implants by co-implantation with Ga⁺. The other objective was to look into the possibility of controlling the degree of compensation by co-implantation of As⁺ and possibly forming n-type layers. The amphoteric nature of C⁺ in GaAs allows co-implantation by forcing C⁺ into arsenic or gallium sites. Perhaps the most convincing data obtained were for samples implanted with 10¹⁴ dose and annealed at 700°C as shown in Table III. In the table, the resistivity of samples implanted with C⁺ is lower by two orders of magnitude. Co-implantation of Ga⁺ enhances the doping efficiency while As⁺ co-implantation reduces the electrical activity. The encapsulation used for the present study appears to be superior to SiO₂ layers used by Sansbury or Harris (Ref 8,9) as can be seen from the table.

Table III
Electrical Characteristics of 10¹⁴
dose and 700°C annealed samples

Sample	β _S (Ω/ □)	μ _{eff} (cm ² /V-sec)	P _s (.cm ⁻²)
Encapsulation	1.08 x 10 ⁷		
GaAs:Ar ⁺	5.01 x 10 ⁶		
GaAs:C+	1.17 x 10 ⁴	150	3.57 x 10 ¹²
GaAs:Ga ⁺ , C ⁺	4.56 x 10 ³	114	1.20 x 10 ¹³
GaAs:As ⁺ , C ⁺	1.47 x 10 ⁴	162	2.63×10^{12}
Sansbury	2.72 x 10 ⁴	242	9.50×10^{11}

In addition, other supporting results are shown in the $10^{14}/\text{cm}^2$, 900°C annealed, GaAs:Ga⁺, C⁺, GaAs:As⁺, C⁺, and GaAs:C⁺ samples. Efficiency is highest for the Ga⁺ co-implantation with a value of 32%. As gave an efficiency of 25% while C alone was only 9.5%. Although the As⁺, C⁺ co-implantation was expected to be lower than the C⁺ alone, the gross damage effect due to the heavy ions seems to aid the efficiency obtained in these samples at this annealing temperature. The annealing behavior discussed so far can be more clearly observed from the profiles obtained. The $GaAs:As^+$, C^+ sample of $10^{12}/cm^2$ dose, annealed at 900°C (Fig. 12) is most peculiar of them ail, since the concentration profile is substantially higher than either LSS profile or sublattice vacancy profile obtained from Pearson and Chiang (Ref 10). This phenomena suggest, as stated previously, the encapsulation may act as a sink and enhance the out-diffusion of the sublattice atoms. It would account for the high p-type concentration at the surface. This observation is reinforced by comparing the concentration profile to the

gallium vacancy profile, which changes as a complementary error function. The concentration profiles appear to be independent of dose or type of implants as can be seen by comparing Fig. 12 through 15. This indicates that it is very difficult to separate the effects due to encapsulation from those due to implantation using the profiles. In spite of this difficulty, it is still interesting to compare the $10^{14}/\text{cm}^2$, 900°C samples of GaAs:As⁺, C⁺ and GaAs:Ga⁺, C⁺. It can be observed from Fig. 14 and 15 that Ga⁺ co-implantation gives better profile integrity. This can be seen from the low compensation coefficient near the Ga⁺ implanted area. On the other hand the As⁺ co-implantation gives a broad profile with no sharp average change in compensation with depth. The high bulk concentration $(10^{19}/\text{cm}^3)$ near the surface in Fig. 14 might be due to damage enhanced doping effect as indicated by the high compensation value in this region.

V. Summary, Conclusion and Recommendations

This thesis looked at the possibility of co-implantation of Ast and Gat with Ct, in order to reduce the electrical activity of the dopant in one case, and to increase in the other. GaAs was doped with C^{\dagger} alone and with equal doses of As^{\dagger} and Ga^{\dagger} to maintain the stoichiometry of the crystal. The samples were all encapsulated with the same material and thickness, and annealed for the same period of time. The samples implanted with C⁺ only were done to verify Shin's results and used as a reference for the co-implantation samples. Only surface measurements were made on most of the samples due to the extended amount of time required to profile each one. The only ones that were profiled were the 900°C annealed samples. Only five profiles are presented because it was not possible to obtain any Hall voltages on the others in addition, the etch rate was so fast that after two etches the implanted region was passed. The samples for which no profiles were obtained could have been run aga-n but the ion implantation facility was not operational, and therefore could not produce more samples. From the data obtained, it is seen that some of the predicted results were verified.

The results presented in this thesis have shown to some extent that the addition of Ga⁺ or As⁺ implants to the primary implants C⁺, has an effect on the doping efficiency and crystal lattice. This effect is due to the encapsulation, which may be acting as a sink for out-diffusion of gallium vacancies, the enhancement of site location due to the Ga⁺, As⁺ implants, and possibly the damage of the sublattice. All these effects contribute to the efficiency which was observed, but they must be studied separately to see how much each one contributes.

Additional work should be carried out in co-implantation for controlling the stoichiometry of the GaAs, which shows some promise for ion implantated layers. More measurements should be acquired to show the reproducibility of this process. The encapsulation which has plagued this work should be more thoroughly researched as it hinders the observation of implantation results. The damages and reactions produced by the implantation of Ga⁺ and As⁺ should also be looked into. Temperature dependence measurements should be made to better distinguish the effects which are dominant in the conduction process. Different annealing schemes should also be tried. For example, annealing could be done after Ga⁺ or As⁺ implantation, followed by implantation of the primary species with post annealing. This may relieve some of the damage due to the heavy ion implantation.

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Appendix A

Definition of Symbols

F(g)	a foreign atom in the gas phase
FGA, FAS	a foreign atom on a Ga or As lattice site in GaAs
F _{si}	a foreign atom on a Si lattice site
h	a hole in the valence band
e .	an electron in the conduction band
x, 'and'	supperscripts denote neutral, single positively charged and single negative charged species respectively
VAS. VGA	a vacant site in the arsenic or gallium sublattice
Ga _i , As _i	a gallium or arsenic atom on an interstitial site
	(<u>Ref 1</u>)

Appendix B

			Flow rates	Flow rates for deposition of $\mathrm{Si}_{3}\mathrm{N}_{4}$	ition of :	Si3N4		,		-	
	Film	Sign (Sech)	(ŠCČM)	NH3 (SÇÇH)	(SCEM)	Set Temp	T.C. nv	Ţime	Hold Temp	Before AFTE	LN ₂
Gat, ct	SiaMa	80	500	200	500	725	29.7	30 sec	200°C	B/A	LN2
	Si3NA	80	200	200	200	725	29.7	30 sec	200°C	B/Ä	LN,
	S1314	80	200	200	200	725	29.7	30 sec	200°C	B/A	LN ₂
Ast,ct	Si3N4	80	200	200	200	725	29.8	30 sec	200°C		LN ₂
	Siand	80	200	. 200	200	725	29.8	30 sec	200°C		LN,
	Si3N4	80	200	200	200	725	29.8	30 sec	200°C		LN2 2
+ 5	Si3114	80	. 200	, 200	200	725	29.8	30 sec	200°C	B/A	LN2
	Si 314	80	200	200	200	725	29.8	30 sec	200°C	B/A	LN
	Si3N4	80	200	200	200	725	29.8	30 sec	200°C	B/A	LN ₂

Appendix C

Order	Color	SiO, Thickness Range* (µm)	Si ₂ N ₄ Thickness Range (µm)
	Silicon	0-0,027	0-0,020
	Brown	0.027-0:053	0.020-0.010
	Golden Brown	0,053 0,073	0.040-0.055
	Red	0,073 -0,097	0,055-0,073
	Deep Blue	0.097-0.010	0.073-0.077
ist	Blue	0.10~0.12	0,077-0,093
	Pale Blue	0.120.13	0,093-0,10
	Very Pale Blue	0,13-0,15	0.10-0.11
	Siticon	0,15-0,16	0,11 0,12
	Light Yellow	0,16~0,17	0.12-0.13
	Yellow	0,17-0,20	0.13-0.15
	Orange Red	0.20~0.24	0.15 0.18
lst	Red	0.24-0,25	0,18 0,19
	Dark Red	0.25-0.28	0,19 0,21
2nd	Blue	0.28-0.31	0.21 -0.23
	Blue-Green	0.31-0.33	0,23-0,25
	Light Green	0.33~0.37	0.25 -0.28
	Orange Yellow	. 0.37-0.40	0.28-0.30
2nd	Red	0,40-0,44	0.30 0.33

Fig. C.1 Color corresponding to varying thicknesses of the deposited layer

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Appendix D

Table IV
GaAs:C⁺

p(cm ⁻²)	T _A (°C)	ρ _s (Ω/ロ)	^μ eff(cm ² /V-sec)	P _s (cm ⁻²)
10 ¹²	600 700 800 900	1.35 x 10 ⁸ 2.78 x 10 ³ 2.47 x 10 ³	160 101	1.41 x 10 ¹² . 2.50 x 10 ¹³
10 ¹³	600	2.34 x 10 ⁴	176	1.52 x 1012
	700	3.02 x 10 ⁴	153	1.35 x 1012
	800	1.32 x 10 ³	161	2.94 x 1012
	900	5.90 x 10 ³	185	5.73 x 10
10 ¹⁴	600	8.07 x 10 ³	108	7.14 x 1012
	700	1.17 x 10 ³	150	3.57 x 1012
	800	8.43 x 10 ³	148	5.00 x 1012
	900	4.58 x 10 ³	143	9.50 x 1012

Table V
GaAs:Ga⁺, C⁺

Ø(cm ⁻²)	T _A (°C)	Γ _S (Ω/ D)	μ _{eff} (cm²/V-sec)	P _s (cm ⁻²)
1012	600 700 800 900	\sim_{10}^{10} 2.14 x 10 ⁹ 9.92 x 10 ³ 2.10 x 10 ³	138	4.50 x 10 ¹²
10 ¹³	600 700 800 900	4.05 x 104 6.84 x 103 5.71 x 103 1.20 x 10	126 70 125	1.23 x 1012 1.30 x 1012 8.70 x 1012
10 ¹⁴	600 700 800 900	4.21 x 10 ³ 4.56 x 10 ³ 2.48 x 10 ³ 1.67 x 10 ³	95 114 88 116	1.56 x 10 ¹³ 1.20 x 10 ¹³ 2.80 x 10 ¹³ 3.32 x 10 ¹³

Table VI
GaAs: As⁺, C⁺

Ø(cm ⁻²)	T _A (°C)	β _S (Ω/□)	μ eff ^{(cm²} /V-sec)	P _s (cm ⁻²)
10 ¹²	600 700 800 900	3.10 x 10 ⁹ 4.60 x 10 ⁴ 3.26 x 10 ³ 2.50 x 10	77 150	2.50 x 10 ¹² 1.66 x 10 ¹³
10 ¹³	600 700 800 900	4.95 x 104 9.41 x 103 9.97 x 103 3.31 x 10		9.43 x 1011 6.58 x 1012 5.00 x 1013 1.66 x 10
10 ¹⁴	600 700 800 900	1.55×10^{4} 1.47×10^{4} 1.59×10^{3}	145 162 158	2.77×10^{12} 2.63×10^{13} 2.50×10^{13}

Table VII
Encapsulation

T _A (°C)	· β _s (Ω/□)
600	9.22 x 10 ⁷ 1.08 x 10 ₃ 6.02 x 10 ₃ 3.20 x 10 ³
700	$1.08 \times 10^{\prime}_{3}$
800	6.02×10^{3}
900	3.20 x 10 ³

$$n_{\text{mas}} (R_p) = \frac{\rho}{\Delta R_p \sqrt{2\pi}}$$
 (B.2)

Using these equations and range tables, the theoretical profiles can be determined. For this thesis range tables could not be found for the implantation of As⁺ and Ga⁺ into GaAs, but a computer program which was available at AFAL, produced the tables required.

The distribution can be tailored for the required needs by coating the surface with some substance which has a stopping power close to that of the substrate. Therefore the peak is moved closer to the surface by a distance equal to the coating. Then you would have a higher concentration at the surface and one junction instead of two (see Fig. B.1).

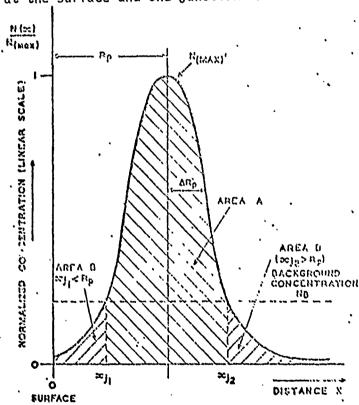


Fig. 5.5. A typical theoretical Gaussian ion distribution with depth, R_n is the projected range, AR_p the standard deviation and N_0 the background doping concentration, X_{I_1} and X_{I_2} are the positions of p-n junctions if type conversion is produced. (See text for explanation of faces A and B.)

Fig. B.1 Theoretical Gaussian implant profile (Ref 13)

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Appendix F

Range-Profiles of Ion Implanted Samples

The Linhard, Scharff and Schiott (LSS) theory was developed on the bases that the implant ions were accelerated into a amorphous target. The reason for this line of thought is that the theory for range distribution in crystals is not well developed (Ref 12). To overcome this problem the crystal is oriented off axis when being implanted or ordered from the manufacture grown in a specific direction. Therefore, the implanted ions will not channel and the substrate will look more amorphous in nature. The GaAs substrates used in this thesis were implanted 7° off the axis of the [100] direction. The range distribution will depend on the energy, mass and nuclear charge of the ion, as well as the mass and density of the substrate. The combination of these parameters results in a Gaussian distribution which has some range and standard deviation (Fig. B.1).

The impurity density from the surface can be represented by the following equation:

$$n(\lambda) = \frac{0}{\Delta R_p \sqrt{2\pi}} \exp \left[\frac{-(x - R_p)^2}{2 \Delta R_p^2} \right]$$
 (B.1)

where x = depth from surface,

 \emptyset = ion dose/cm²,

 ΔR_{p} = standard deviation in projected range,

and $R_n = projected range.$

The peak impurity concentration can be determined by setting \boldsymbol{x} equal to \boldsymbol{R}_n :

$$n_{\text{mas}} (R_p) = \frac{p}{\Delta R_p \sqrt{2\pi}}$$
 (B.2)

Using these equations and range tables, the theoretical profiles can be determined. For this thesis range tables could not be found for the implantation of As⁺ and Ga⁺ into GaAs, but a computer program which was available at AFAL, produced the tables required.

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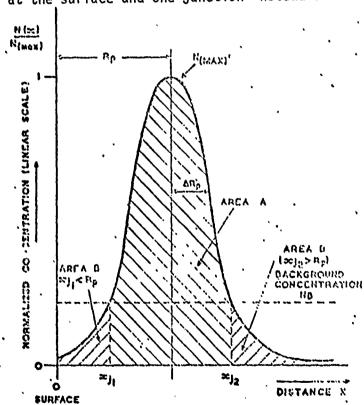


Fig. 5.5. A typical theoretical Gaussian ion distribution with depth. R_n is the projected range, AR_n the standard deviation and N_0 the background doping concentration. X_{I_1} and X_{I_2} are the positions of p-n junctions if type conversion is produced. (See text for explanations of the set A and B.)

Fig. B.1 Theoretical Gaussian implant profile (Ref 13)

ATIV

Michael Stefiniw was born August 23, 1954 in Passaic, New Jersey. He graduated from Morris Catholic High School in 1972 and attended Stevens Institute of Technology in Hoboken, New Jersey. He received his commission from ROTC in May 1976. His first assignment was the Air Force Institute of Technology, Wright Patterson Air Force Base, Ohio, where he is enrolled in the Graduate Electro-Optics program.

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